固化对水性环氧粘结固体 润滑涂层理化及摩擦学性能的影响

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[摘 要] 为了确定水性环氧粘结固体润滑涂层的最佳固化条件,采用 MFT-R 4000 型往复摩擦磨损试验仪评价 了不同固化条件对其摩擦学性能的影响,用傅立叶变换红外光谱 (FTIR)和示差扫描量热法 (DSC)表征手段,确定 了体系的最佳固化条件。结果表明:固化条件不同,涂层的摩擦学性能差异很大;水性粘结剂与水性固化剂的最佳 质量比为2 :1,最佳固化温度为75 ℃,最佳固化时间为2 h;以最佳条件制备的水性环氧粘结固体润滑涂层具有优 异的理化性能和摩擦学性能,摩擦磨损寿命比传统的有机溶剂型粘结固体润滑涂层约长 30%。本研究为制备高性 能环保型粘结固体润滑涂层提供了可能。

[关键词] 固体润滑涂层;水性环氧树脂;粘结固体润滑剂;固化条件;摩擦学性能 [中图分类号] TQ323.5;TH117.3 [文献标识码] A [文章编号] 1001-1560 (2010)11-0007-03

0 前 言

水性环氧粘结固体润滑剂^[1,2]以水性环氧涂 料^[3~6]为粘结剂,二硫化钼、石墨等为润滑填料、水为 分散介质,不含挥发性有机化合物(VOC),价格低廉、 无气味、不燃、环保等。目前,大多的研究集中在溶剂 型固体润滑剂^[7~9],对水性环氧粘结固体润滑剂的研 究只有少数的专利^[1,2]。水性润滑剂代表着未来的发 展方向。

然而,在水性粘结固体润滑剂的固化过程中,粘结 剂(即水性环氧树脂涂料)容易产生核壳结构,导致固 化不完全,致使涂层性能变差^[10~12];而且与润滑剂的 搅拌混合,也会使其固化性能受到一定的影响。因此, 最佳固化条件的确定对于开发高性能的水性粘结固体 润滑涂层具有非常重要的意义。本工作制备了水性 MoS₂ 粘接润滑涂层,采用自制的往复摩擦磨损试验仪 评价了不同固化条件对水性环氧粘结固体润滑剂摩擦

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学性能的影响;以现代表征技术研究了水性环氧粘结 固体润滑涂层的最佳固化条件;进一步考察了水性环 氧树脂粘结固体润滑涂层最佳固化条件下的理化性能 和摩擦学性能,以期为制备高性能环保型粘结固体润 滑涂层提供可能。

1 试 验

1.1 涂层制备

基材处理:基体材料是 45 钢和马口铁,尺寸均为 12.7 mm ×12.7 mm ×19.0 mm;作常规喷砂、丙酮超声 等处理,除去锈、油。

水性粘结固体润滑涂料配制:将 25.0%固体润滑 剂 (二硫化钼,固含量 99%)、0.5%表面活性剂(聚乙 二醇)和水球磨 48 h;加入一定量的粘结剂(乳白色水 性环氧树脂乳液,固含量 50%)和固化剂(水性黄色黏 稠液聚酰胺,固含量 60%)(两者质量比为 1.0 4.2), 搅拌均匀。

水性环氧粘接固体润滑涂层喷涂:以 0.15~0.20 MPa 压缩氮气作驱动气体,将配制好的涂料喷涂到基材表面;以不同温度进行固化,得到水性 MoS₂粘结润滑涂层,厚度为 (20 \pm) μ m。

1.2 涂层表征

采用 MFT-R 4000 型往复摩擦磨损试验仪评价涂

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层的摩擦学性能:上试样为 AISI 440C 钢球, ∮12.00 mm, 硬度 58 HRC,下试样为具有涂层的 45 钢,两者为 点接触;施加的载荷为 30 N,摩擦长度 2.5 mm,频率 10 Hz,温度 24 ~30 ℃,空气相对湿度 30 % ~50 %。

采用 IFS66V/S 型红外光谱仪 (FTIR)测试涂层固 化后环氧特征峰的变化:将马口铁上的固化涂层用刀 片刮下,用 KBr 粉末压片后进行红外测试。

采用 STA 449C 同步热分析仪 (DSC)测试涂料的 最佳固化温度; 预先将待测试样真空干燥 (室温, 真空 度 0.07 MPa)0.5 h,除去水分;每次约取5 mg,分别以 5,10,15 [℃]/min 升温速率进行测试(以空气作吹扫气, 氮气作保护气)。

分别按 GB 1720 – 89, GB/T 1732 – 93 和 GB 1731 – 89 测试涂层的附着力、耐冲击性和柔韧性^[13]。

2 结果与讨论

2.1 水性粘结剂和水性固化剂最佳配比

5 种水性粘结剂和水性固化剂配比 100 [℃]固化涂 层的摩擦磨损性能见图 1 和图 2。



图 1 不同配比下水性 MoS₂ 粘结润滑涂层的耐磨寿命



图 2 不同配比下水性 MoS₂ 粘结润滑涂层的摩擦系数

从图1,图2可以看出,两者的配比不同,涂层的摩 擦学性能明显不同:当配比为10 5时,涂层的耐磨寿 命(磨穿某一厚度涂层所用时间)较长、摩擦系数最低; 配比大于10 7或者小于10 5时,涂层的耐磨寿命 明显降低。因为涂层的耐磨性与粘结剂的固化程度直 接相关,固化程度越高,交联密度越大,耐磨性就越好, 当水性粘接剂与水性固化剂配比为10 5时,涂层就 完全固化。

涂层的固化交联程度见图3。



图 3 4 种水性粘接剂与水性固化剂配比时涂层的红外光谱

从图 3 可以看出: 1 508 cm⁻¹处的强吸收峰为苯环 特征吸收峰,且不同的配比时保持不变,说明苯环未参 与固化反应; 915 cm⁻¹处涂层的环氧特征峰强度随配 比不同发生了明显的变化,配比为 10 ·1 时,环氧特征 峰较强,说明固化后环氧基有剩余,配比为 10 ·3 时, 环氧特征峰明显减弱,配比为 10 5 时,环氧特征峰消 失,环氧基完全与酰胺基发生如下反应:

$$\begin{array}{c} + \begin{array}{c} C - R - C - NH - R - NH \\ \parallel \\ 0 \end{array} \xrightarrow{+} \begin{array}{c} + \end{array} \xrightarrow{+} \begin{array}{c} - R' \\ - R' \end{array} \xrightarrow{+} \begin{array}{c} - P \\ - R' \\ 0 \end{array} \xrightarrow{+} \begin{array}{c} - P \\ - R - C - N - R - NH \\ 0 \end{array} \xrightarrow{+} \begin{array}{c} - P \\ - R' \\ - CH \end{array} \xrightarrow{+} \begin{array}{c} 0 \\ - R' \\ - CH \end{array}$$

涂层固化交联不完全, 会导致其耐磨性变差。结 合涂层的红外谱 (图 3)及耐磨性数据 (图 1)可以看出, 当两者配比为 10 5 时, 水性 MoS₂ 粘结固体润滑涂层 具有优异的摩擦学性能。

2.2 涂层最佳固化温度

选定水性粘接剂和水性固化剂的配比为10 5,去 除水分后,3 种升温速率对涂层固化的影响见图 4。由 图4 可以看出,放热峰的峰值随升温速率的增加而升 高。固化反应是放热反应,反应越完全,放出的热量越 多。以峰值对升温速率作图,外推升温速率为0 ℃/min 时^[14,15],峰值温度为 75 ℃(见图 5),即在接近于零升 温速率下,75 ℃时放出的热量最多,固化反应最容易发 生。因此,最佳固化温度确定为 75 ℃。



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图 5 峰值对升温速率的变化曲线

2.3 涂层最佳固化时间

选取上述优化条件,将涂层于室温下放置 1 h 后, 75 [°]C固化 1,2,3 h,其耐磨结果见图 6。从图6 可以看 出,固化 1 h,耐磨寿命为 80 min,固化 2 h 耐磨寿命增 长至 120 min,固化 3 h 和 2 h 相近,说明固化 2 h,固化 交联已比较完全。涂层不同时间固化后的红外光谱见 图 7:915 cm⁻¹处室温放置 1 h 时涂层存在明显的环氧 特征峰,环氧基没有完全参与反应,固化交联很不完 全;75 [°]C固化 1 h 后,存在微弱的环氧特征峰,其强度 明显减弱,但仍有未反应的环氧基存在;固化 2 h 后,环 氧峰完全消失,表明环氧基与酰胺基完全发生反应,固 化交联完全。结合涂层的摩擦学和红外数据,确定最 佳的固化时间为 2 h。



图 6 不同时间固化后涂层的耐磨寿命



图 7 不同时间固化后涂层的红外光谱

2.4 涂层理化性能和摩擦学性能

以最佳条件制备的涂层附着力为1级,抗冲击性能 为 60 cm,柔韧性为1 mm。这表明水性环氧树脂润滑涂 层的理化性能可与有机溶剂型润滑涂层相媲美。

水性和有机溶剂型固体润滑涂层的摩擦学性能分

别见图 8 和图 9 (其中不同符号的曲线代表相同条件下 3 次重复试验的结果)。对比图 8 和图 9 可知:水性粘 结固体润滑涂层的磨损寿命为 400 min,比有机溶剂型 的约长 30%,摩擦系数平均值为 0.068,比有机溶剂型 的更低。这为开发高性能环保型的粘结固体润滑涂层 提供了重要参考。



图 9 有机型粘结剂固体润滑涂层的摩擦曲线

3 结 论

(1)水性 MoS₂粘结固体润滑涂层的最佳固化条
 件:水性粘接剂与水性固化剂的最佳配比为 2 · 1,最佳
 固化温度为 75 ℃,固化时间为 2 h。

(2)以最佳固化条件制备的水性粘结固体润滑涂 层主要含二硫化钼、石墨、三氧化二锑等,其理化性能 优异,摩擦系数低、耐磨寿命长,优于有机溶剂型粘结 固体润滑涂层。

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Contents & Abstracts

Oxygen Evolution Electrocatalytic Performance of MnO_2 Coating Anode Electrodeposited on Titanium in Manganese Acetate Electrolyte

GAO Hong, ZHU Cheng-fei, XIE Rui, WANG Xiao-jun (College of Materials Science & Egineering, Nanjing University of Technology, Nanjing 210009, China). Cailiao Bachu 2010, 43 (11), 01 ~03 (Ch). Electrodeposited MnO₂ coatings were prepared on the surface of Ti alloy from the electrolyte consisting of MnSO₄ and the electrolyte consisting of MnO₂ coatings was confirmed by means of X-ray diffraction, and the morphology was observed using a scanning electron microscope. Moreover, the electrochemical performance of the MnO₂ coating anode was evaluated based on cyclic valtammetry, polarization curve test and electrochemical impedance spectroscopy. Results show that the MnO₂ coating prepared from the electrolyte consisting of Mn (CH₃COO)₂ has a faster deposition rate, and as – deposited Ti/MnO₂ anode has better oxygen evolution electrocatalytic performance.

Key words: electrodeposition; MnO_2 coating; crystal structure; Ti alloy; anode; oxygen evolution electrocatalytic performance

Removal of Dissolved Aluminum in Chemical Milling Fluids by Dilute Solution Crystallization

ZHOU Guo-hua¹, FENG Qiang², YI N Mao-sheng¹, WEI Li-an² (1. Chengdu Airplane Industry Group Company Limited China Aviation Industry Complex, Chengdu 610091, China; 2. College of Environmental and Chemical Engineering, Nanchang University of Aero-nautics, Nanchang 330063, China). *Cailiao Baohu* 2010, 43 (11), 04 ~06 (Ch). In view of the treatment and recycle of abandoned chemical milling fluids as well as removal of dissolved Al therein, dilute solution crystallization method based on crystallization and nucleation of sodium aluminate, was adopted to remove the dissolved Al in the abandoned chemical milling fluids. The effects of dilution multiple and temperature on dissolved Al and other components were investigated. Results show that the best dilution multiple is 1.0 ~2.0. At 25 °C it took about 15 d to satisfactorily remove dissolved Al at a rate of about 50% ~60%, while at 55 °C it took about 5 ~6 d to remove the dissolved Al at a rate of 30% ~40%. Moreover, the content of other components such as NaOH, $\rm Na_2S$ and $\rm C_6H_{15}NO_3$ remained unchanged after removing dissolved Al from the abandoned chemical milling fluids via dilute solution crystallization route.

Key words: chemical milling fluid; dilute solution crystallization; regeneration of chemical milling fluid; sodium aluminate; aluminum hydroxide

Influence of Curing on Physicochemical and Tribological Properties of Waterborne Epoxy Resin-Based Bonded Solid Lubricating Coatings

WANG Yue-mei^{1, 2}, ZHOU Hui-di¹, CHEN Jian-min¹, CHEN Lei¹, YE Yin-ping¹ (1. State Key Laboratory of Solid Lubrication, Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences Lanzhou 730000, China; 2. Graduate University, Chinese Academy of Sciences, Beijing 100049, China). *Cailiao Baohu* 2010, 43 (11), 07 ~09 (Ch). An MFT-R 4000 reciprocating friction and wear tester was performed to investigate the effect of curing conditions on the tribological properties of waterborne epoxy resin-based bonded solid lubricating coatings. The optimal curing conditions were established by combining Fourier transformation infrared (FTIR) spectrometry with differential scanning calorimetry (DSC). It was found that waterborne epoxy resin-based bonded solid lubricating coatings obtained under different curing conditions possessed different tribological properties.

The optimal mass ratio of the epoxy resin emulsion to curing agent was determined to be 2 \cdot 1, while the optimal curing temperature and time were suggested as 75 $^{\circ}C$ and 2 h respectively. Resultant waterborne epoxy resin-based bonded solid lubricating coating showed excellent physicochemical and tribological properties, and the antiwear life was longer than that of traditional solvent-based bonded solid lubricating coating by 30%.

Key words: solid lubricating coating; waterborne epoxy resin; bonded solid lubricant; curing condition; tribological property

Application of Stereomicroscope in Study of Corrosion of Aluminum Alloy

WANG Yu- ya^{1, 2}, HAN En - hou¹ (1. Institute of Metal Research, Chinese Academy of Sciences, Shenyang 110016, China; 2. No. 10 Laboratory, Beijing Research Center of Aeronautical Technology, Beijing 100076, China). Cailiao Baohu 2010, 43 (11), 10 ~13 (Ch). With the assistance of stereomicroscopic analysis attempts were made to realize real time observation and recurring of the corrosion process of LY12CZ Al alloy. The application of stereomicroscope in corrosion tests was explored. It was found that in EXCO solution the corrosion of Al alloy was initiated along with the destruction of surface passivation film. Corroded pits were extended along the rolling direction, which was still true even in the presence of artificial scratches perpendicular to the rolling direction. Besides, unstable colloidal Al (OH), film was able to hinder the development of local corrosion, and the regions where corrosion was initiated in preliminary stage were not necessarily to experience severe corrosion. It was feasible to realize real time observation and recurring surface corrosion process by making use of stereomicroscope and relevant technique. It is possible to more accurately study corrosion process by combining macroscopic observation with microcosmic one if limits in relation to magnification times are overcome.

Key words: LY12CZ Al alloy; corrosion; stereomicroscope; macroscopic; microcosmic

Formation Mechanism of Powder-Like Rust and Its Influence on Corrosion of Bronze Wares Studied by Electrochemical Methods

FENG Li-ting, SU Chang, FENG Shao-bin, LI Zhen-xing, WANG Shao-liang (Henan Provincial Key Laboratory of Surface & Interface Science, Department of Material and Chemical Engineering, Zhengzhou University of Light Industry, Zhengzhou 450002, China). Cailiao Bachu 2010, 43 (11), 14 ~16 (Ch). Several kinds of insoluble nantokite were used as active catalysts to prepare porous oxygen electrodes. Resultant porous oxygen electrodes were combined with bronze anode to produce primary batteries. The primary batteries were continuously discharged at a current density of 0.01 mA/cm² for more than 115 h, and the cathodic polarization of the porous oxygen electrodes was tested. The catalytic mechanism of the porous oxygen electrodes was discussed. It was found that the discharge test could be well used to simulate the electrochemical corrosion process of bronze wares under atmosphere. Cuprous chloride rather than basic cupric chloride was the nantokite with the highest catalytic activity for the corrosion process of bronze wares, and the powder-like rust was just the final corroded product.

Key words: corrosion of bronze wares; powder-like rust; simulation of primary batteries; porous oxygen electrodes; cuprous chloride

Corrosion Inhibition Behavior of Benzotriazole for 1Cr18Ni9T i as Nuclear Decommissioning Steel in Chemical Cleaning Agent GONG Min¹, WEI Wei¹, ZHANG Yu¹, KANG Wu², ZHANG